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CHAPTER 2 PRACTICE TEST

Algebra II Chapter 2 Practice Test Name: Do problems by hand, showing work unless otherwise instructed. Calculators are OK.1. For the points A = (5, -4) and B = (1, -12) a) Find the midpoint of AB b) Find the length of AB c) Find the slope of AB d) Find the equation of the line through A and B e) Find the coordinates of point C such that A is the midpoint of BC f) Honors Extra Credit: What point ...

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MICROSOFT WORD - CHAPTER P PRACTICE TEST.DOC

Algebra 2 P Chapter Practice Test Name: Please show your work. You are allowed only a basic calculator. This is last years test.1. Perform the indicated operations and simplify 2a) $(5x^2 - 7y)$ b) $(2x + 7)(3x^2 - 8x + 1)^2$. Factor each expression completely a) $x^2 - 3x - 40$ b) $2x^3 + x^2 - 6x - 3c) x^4 + 5x^3 + 6x^2 d) 2(a + b)^2 + 5(a + b) - 33$. a) Write $3 < x < 8$ in interval notation and...

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UNIT 4 PRACTICE TEST

Honour Chemistry: Unit 4 Practice Test: Thermochemistry and Nuclear ChemistryPart A: Multiple Choice and Numerical Response (1 mark each for Multiple Choice) (2 marks each for Numerical Response) $E_{\text{bind}} = ??mc^2$
 $c = \text{speed of light } (3.00 \times 10^8 \text{ m/s})$ $1 \text{ eV} = 1.69 \times 10^{-19} \text{ J}$ $(1 \text{ kg} = 6.022 \times 10^{26} \text{ amu}, \text{ or } 1 \text{ g} = 6.022 \times 10^{23} \text{ amu} = 1 \text{ mole amu})$ $1 \text{ MeV} = 1.69 \times 10^{-16} \text{ J}$ $(1 \text{ Neutron} = 1.008665 \text{ amu})$ $(1 \text{ Proton} = 1.007825 \text{ am...})$

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UNIT 2 (PART 1) PRACTICE TEST

Honour Chemistry Practice Test: Unit 2 (Part 1): Matter as Solutions and Gases $1 \text{ atm} = 760 \text{ mm Hg} = 760 \text{ torr} = 101.325 \text{ kPa}$ $TK = TC + 273.15$ $STP = 22.4 \text{ L/mol @ } 0.00C \text{ \& } 101.325 \text{ kPa (1 atm)}$ $SATP = 24.8 \text{ L/mol @ } 25.0C \text{ \& } 100. \text{ kPa}$ $L \text{ atm}$ $L \text{ kPa}$ $m \text{ m}$ $n \text{ solute}$ $R = 0.0821 \text{ or } 8.314$ $D = n = C = K \text{ mol}$ $K \text{ mol}$ $V \text{ M}$ $V \text{ solution}$ $P1V1 = P2V2$ $r1$ $M2 n1$ $P1 C1V1 = C2V2$ $PV = nRT = = ?1 = = n1T1 n2T2 r2$ $M1 n \text{ Total}$ $PT \text{ Total}$ $PT \text{ Total} = P1 + \dots$





















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